

Midterm Review or: Everything you ever wanted to know about the Hydrogen atom (but were afraid of coming to my O.H.s to ask)

Spectrum: $E_n = - \left[\frac{me^4}{2\hbar^2} \right] \frac{1}{n^2} = -13.6 \text{ eV} \left(\frac{1}{n^2} \right) \quad n=1, 2, 3, \dots$

Ground State, $E_1 = -13.6 \text{ eV}$ (I've written all this in CGS units - you can recover MKS by just making the substitution $e^2 \rightarrow \frac{e^2}{4\pi\epsilon_0}$ everywhere). Recall that:

SI Units	Gaussian (CGS)	$\hbar = 1.054 \times 10^{-34} \text{ J}\cdot\text{s} = 1.054 \times 10^{-27} \text{ erg}\cdot\text{s}$
1 m	= 100 cm	$\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \frac{\text{Nm}^2}{\text{C}^2} \quad k = \frac{e^2}{\hbar c} \approx \frac{1}{137}$
1 kg	= 1000 g	$e = 4.803 \times 10^{-10} \text{ esu}$
1 N	= 10^5 dyne	$= 1.602 \times 10^{-19} \text{ C}$
1 J	= 10^7 erg	$1 \text{ eV} = 1.602 \times 10^{-12} \text{ erg} = 1.602 \times 10^{-19} \text{ J}$
1 Coulomb	= $3 \times 10^9 \text{ esu}$	$a_0 = \frac{\hbar^2}{me^2} = 0.529 \text{ \AA} = 5.29 \times 10^{-9} \text{ cm}$
1 Ampere	= $3 \times 10^9 \text{ esu/sec}$	$E_1 = -\frac{e^2}{2a_0}$
1 V/m	= $\frac{1}{3} \times 10^{-4} \text{ statvolt/cm}$	$m_e = 9.11 \times 10^{-31} \text{ kg} = 9.11 \times 10^{-28} \text{ g}$
1 Tesla	= 10^4 gauss	$m_p = 1.67 \times 10^{-27} \text{ kg} = 1.67 \times 10^{-24} \text{ g}$
1 eV	=	$R = \text{Rydberg Constant} = \frac{me^4}{4\hbar^2} = 1.097 \times 10^7 \text{ m}^{-1}$

These energy levels enter most directly into physically observable quantities in experiments observing the emission spectrum of excited Hydrogen. The energies of the emitted photons correspond to the difference in energy between initial & final states:

$E_\gamma = E_i - E_f = -13.6 \text{ eV} \left(\frac{1}{n_i^2} - \frac{1}{n_f^2} \right)$
 ~~$n_i = 2, 3, \dots$~~ $n_i = 2, 3, \dots$
 ~~$n_f = 1, 2, 3, \dots$~~ $n_f = 1, 2, 3, \dots$

Now, $E_\gamma = h\nu$ (Planck's Law), $\lambda = c/\nu \Rightarrow \frac{1}{\lambda} = R \left(\frac{1}{n_f^2} - \frac{1}{n_i^2} \right)$, $R = 1.097 \times 10^7 \text{ m}^{-1}$

Lyman Series $n_f = 1$ UV α -line $n_i = 2 \rightarrow n_f = 1$ $n_i \rightarrow \infty \lambda \rightarrow \frac{1}{R} \approx 909 \text{ nm}$

Balmer Series $n_f = 2$ Visible " $n_i = 3 \rightarrow n_f = 2$

Paschen Series $n_f = 3$ IR " $n_i = 4 \rightarrow n_f = 3$

Q: What color is the $n_i = 5 \rightarrow n_f = 2$ transition ^{like} in the Balmer Series?

A: $\frac{1}{\lambda} = R \left(\frac{1}{2^2} - \frac{1}{5^2} \right) = 0.21R \approx 0.23 \times 10^7 \text{ m}^{-1} \Rightarrow \lambda = 434 \text{ nm}$ (Blue!)
 (violet) (red)

The visible range is from: 400 nm - 700 nm

(using $1 \text{ eV} \sim 2.42 \times 10^{14} \text{ Hz}$) $3.1 \text{ eV} - 1.8 \text{ eV}$
 $hc = 1.24 \times 10^3 \text{ eV}\cdot\text{nm}$ $7.64 \times 10^{14} \text{ Hz} - 4.22 \times 10^{14} \text{ Hz}$

Q: Problem 4.17 in Griffiths: Determine the Bohr energies, $E_n(Z)$, the binding energy $E_1(Z)$, the Bohr radius ^{a(2)} & the Rydberg constant $R(Z)$ for a hydrogenic ion of atomic # Z (ignore relativistic & reduced mass effects!)

A: In this approximation, the only change is that the potential: $V(r) = -\frac{e^2}{r} \rightarrow V(r) = -\frac{Ze^2}{r}$. So we get all the corresponding values by making the substitution $e^2 \rightarrow Ze^2$ in all formulas!

$$\begin{aligned}
 E_n(z): \quad E_n(1) = -\frac{1}{n^2} \frac{m e^4}{2 \hbar^2} &\longrightarrow E_n(z) = -\frac{1}{n^2} \frac{m (ze^2)^2}{2 \hbar^2} = -\frac{z^2}{n^2} (13.6 \text{ eV}) \\
 E_1(z): \quad E_1(1) = -13.6 \text{ eV} &\longrightarrow E_1(z) = -z^2 (13.6 \text{ eV}) \\
 a(z): \quad a(1) = a_0 = \frac{\hbar^2}{m e^2} &\longrightarrow a(z) = \frac{\hbar^2}{z m e^2} = \frac{a_0}{z} = 0.529 \times 10^{-10} \text{ m} / z \\
 R(z): \quad R(1) = \frac{m e^4}{4 \pi^2 \hbar^3} &\longrightarrow R(z) = z^2 \frac{m e^4}{4 \pi^2 \hbar^3} = z^2 (1.097 \times 10^7 \text{ m}^{-1})
 \end{aligned}$$

Q: What is the degeneracy of E_5 ?

A: Given a state with principal quantum number n , the allowed values of l are: $l = 0, 1, 2, \dots, n-1$. Further, for each l , there are $(2l+1)$ possible values of m , $m = -l, -l+1, \dots, l-1, l$, so we get:

$$\text{degeneracy of } n^{\text{th}} \text{ state} = \sum_{l=0}^{n-1} (2l+1) = 2 \left(\frac{n(n-1)}{2} \right) + n = n^2$$

Thus, the degeneracy of E_5 is 25. Note: We have actually neglected several important physical effects in the Hydrogen atom (spin, relativistic corrections, Lamb shift) that break much of this degeneracy (I'll talk a little bit about this in section!)

Other useful facts about Hydrogen Atom:

$$\Psi_{n\ell m}(r, \theta, \phi) = R_{n\ell}(r) Y_{\ell}^m(\theta, \phi) \quad R_{n\ell}(r) = \frac{1}{r} \rho^{\ell+1} e^{-\rho} v(\rho)$$

where $\rho = \frac{r}{a_0 n}$. Note that for large r , $R_{n\ell} \sim e^{-\rho} \sim e^{-r/a_0 n}$.

Thus, one can qualitatively argue that the "size" of the orbit scales as n , so $\Psi_{n\ell m}$ has an "effective" radius of $a(n) \sim a_0 n$. For very small r , one finds $R_{n\ell}(r) \sim r^\ell$, so only for $l=0$ does this ~~stay~~ stay non-zero near the origin!

$$\begin{aligned}
 R_{10} &= 2a^{-3/2} e^{-r/a} & R_{20} &= \frac{1}{\sqrt{2}} a^{-3/2} \left(1 - \frac{r}{2a}\right) e^{-r/2a} \\
 \text{etc...} & & R_{21} &= \frac{1}{\sqrt{24}} a^{-3/2} \frac{r}{a} e^{-r/2a}
 \end{aligned}$$

Angular Momentum: $\vec{L} = \vec{r} \times \vec{p}$ $[L_x, L_y] = i\hbar L_z$ Note: Being observed \vec{L} are Hermitian $L_x^2 = L_x^2, L_y^2 = L_y^2, L_z^2 = L_z^2$

$$[L_z, L_x] = i\hbar L_y \quad [L_y, L_z] = i\hbar L_x \quad L^2 = L_x^2 + L_y^2 + L_z^2$$

Q: Prove $[L^2, L_x] = 0$

$$\begin{aligned}
 \text{A: } [L^2, L_x] &= [L_x^2 + L_y^2 + L_z^2, L_x] = [L_x^2, L_x] + [L_y^2, L_x] + [L_z^2, L_x] \\
 &= L_y [L_y, L_x] + [L_y, L_x] L_y + L_z [L_z, L_x] + [L_z, L_x] L_z \\
 &= -i\hbar (L_y L_z + L_z L_y) + i\hbar (L_z L_y + L_y L_z) = 0
 \end{aligned}$$

Thus, we in fact have $[L^2, L_x] = [L^2, L_y] = [L^2, L_z] = 0$.

However, as L_x, L_y, L_z do not commute with each other, we can only consider orthonormal bases of simultaneous eigenvectors of L^2 and one of the three - if we choose L_z , we get the Yem's!

The Yem (θ, ϕ) are a complete set of orthonormal simult. eigenfns of $L^2 + L_z$ with the properties that:

$$\begin{aligned}
 L^2 Y_{\ell m}(\theta, \phi) &= \hbar^2 \ell(\ell+1) Y_{\ell m}(\theta, \phi) & \ell &= 0, 1, 2, \dots \\
 L_z Y_{\ell m}(\theta, \phi) &= m\hbar Y_{\ell m}(\theta, \phi) & m &\in -\ell, -\ell+1, \dots, \ell-1, \ell
 \end{aligned}$$

In the spirit of the SHO problem, I'm going to try to abstractly approach problems involving angular momentum by using operator methods and Dirac bra-ket notation as well as "raising" + "lowering" or ladder operators.

Bra-ket notation: $Y_{\ell m}(\theta, \phi) \rightarrow |\ell m\rangle$ $\int Y_{\ell m}^*(\theta, \phi) d\Omega \rightarrow \langle \ell m |$

so: $\langle \ell m | \ell' m' \rangle = \delta_{\ell \ell'} \delta_{m m'}$ $L^2 |\ell m\rangle = \hbar^2 \ell(\ell+1) |\ell m\rangle$

$L_z |\ell m\rangle = m\hbar |\ell m\rangle$

Ladder Operators: $L_+ = L_x + iL_y$ $L_- = L_x - iL_y$ ($L_+^\dagger = L_-$, $L_-^\dagger = L_+$)

Properties: ① $[L^2, L_\pm] = [L^2, L_x \pm iL_y] = 0$

② $L_\pm L_\mp = L^2 - L_z^2 \pm \hbar L_z$ PF: $L_\pm L_\mp = (L_x \pm iL_y)(L_x \mp iL_y) = L_x^2 + L_y^2 \mp i[L_x, L_y]$
 $= L_x^2 + L_y^2 \pm \hbar L_z = L^2 - L_z^2 \pm \hbar L_z$

③ $[L_+, L_-] = [L_x + iL_y, L_x - iL_y] = -i[L_x, L_y] + i[L_y, L_x] = 2\hbar L_z$

④ $[L_z, L_\pm] = [L_z, L_x \pm iL_y] = i\hbar L_y \pm i(-i\hbar L_x) = \pm \hbar L_\pm$

⑤ $L_x = \frac{1}{2}(L_+ + L_-)$ ⑥ $L_y = \frac{1}{2i}(L_+ - L_-)$

Q: What are the results of applying L^2 + L_z to the state $L_\pm |\ell m\rangle$?

A: ① Since $[L^2, L_\pm] = 0 \Rightarrow L^2(L_\pm |\ell m\rangle) = L_\pm(L^2 |\ell m\rangle) = L_\pm \hbar^2 \ell(\ell+1) |\ell m\rangle$
 $\Rightarrow L^2(L_\pm |\ell m\rangle) = \hbar^2 \ell(\ell+1)(L_\pm |\ell m\rangle)$ Thus, $L_\pm |\ell m\rangle$ has the same value of ℓ as $|\ell m\rangle$ itself.

② $L_z L_\pm |\ell m\rangle = (\text{by prop. ④ above}) = \hbar(L_\pm L_z \pm \hbar L_\pm) |\ell m\rangle$
 $= (m\hbar L_\pm \pm \hbar L_\pm) |\ell m\rangle = (m \pm 1)\hbar L_\pm |\ell m\rangle$

Thus, L_\pm act like ladder operators increasing or decreasing the value of m by ± 1 !

Q: From the result of the above problem, one would guess that $L_\pm |\ell m\rangle = c_{\ell m}^\pm |\ell, m \pm 1\rangle$. Using the fact that $L_+^\dagger = L_-$ show that: $L_\pm |\ell m\rangle = \hbar [\ell(\ell+1) - m(m \pm 1)]^{1/2} |\ell, m \pm 1\rangle$

PF: All we need to do is find: (as $|\ell, m \pm 1\rangle$ is properly normalized!)

$|c_{\ell m}^\pm|^2 = (L_\pm |\ell m\rangle)^\dagger L_\pm |\ell m\rangle = \langle \ell m | (L_\pm)^\dagger L_\pm |\ell m\rangle = \langle \ell m | L_\mp L_\pm |\ell m\rangle$
 $= \langle \ell m | (L^2 - L_z^2 \mp \hbar L_z) |\ell m\rangle = \langle \ell m | (\hbar^2 \ell(\ell+1) - m^2 \hbar^2 \mp \hbar^2 m) |\ell m\rangle$
 $= \hbar^2 [\ell(\ell+1) - m(m \pm 1)] = |c_{\ell m}^\pm|^2$

\Rightarrow After normalizing, we get taking the square root $\Rightarrow c_{\ell m}^\pm = \hbar [\ell(\ell+1) - m(m \pm 1)]^{1/2}$

Q: In the state $\Psi = \frac{1}{\sqrt{2}}(|\ell m\rangle + |\ell, m+1\rangle)$, find $\langle L_x \rangle$, $\langle L_y \rangle$, $\langle L_x^2 \rangle$ + ~~...~~

A: $\langle L_x \rangle = \frac{1}{2} (\langle \ell m | + \langle \ell, m+1 |) (L_x) (|\ell m\rangle + |\ell, m+1\rangle)$
 $= \frac{1}{2} (\langle \ell m | L_x | \ell m \rangle + \langle \ell, m+1 | L_x | \ell m \rangle + \langle \ell m | L_x | \ell, m+1 \rangle + \langle \ell, m+1 | L_x | \ell, m+1 \rangle)$

Note, as $L_x = \frac{1}{2}(L_+ + L_-)$ and as $\langle \ell m | L_\pm | \ell m \rangle = \hbar [\ell(\ell+1) - m(m \pm 1)]^{1/2} \langle \ell m | \ell, m \pm 1 \rangle = 0$
 $\Rightarrow \langle \ell m | L_x | \ell m \rangle = \langle \ell, m+1 | L_x | \ell, m+1 \rangle = 0!$

$\Rightarrow \langle L_x \rangle = \frac{1}{2} (\langle \ell, m+1 | \frac{1}{2}(L_+ + L_-) | \ell m \rangle + \langle \ell m | L_x | \ell, m+1 \rangle)$
 $= \frac{\hbar}{4} ([\ell(\ell+1) - m(m+1)]^{1/2} \langle \ell, m+1 | \ell, m+1 \rangle + [\ell(\ell+1) - m(m-1)]^{1/2} \langle \ell, m+1 | \ell, m-1 \rangle)$
 $+ [\ell(\ell+1) - m(m+1)]^{1/2} \langle \ell m | \ell, m+2 \rangle + [\ell(\ell+1) - m(m+1)]^{1/2} \langle \ell m | \ell m \rangle]$
 $= \frac{\hbar}{2} ([\ell(\ell+1) - m(m+1)]^{1/2})$ The rest is part of review problems!